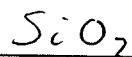


Write the name of the molecular compound from the given formula.

1. NO_2 nitrogen dioxide
2. CO_2 carbon dioxide
3. SO_3 sulfur trioxide
4. P_2S_5 diphosphorus pentasulfide
5. ICl_3 iodine trichloride
6. N_2O_4 dinitrogen tetroxide
7. ICl iodine monochloride
8. SiCl_4 silicon tetra chloride
9. SF_2 sulfur difluoride
10. SF_6 sulfur hexafluoride
11. NCl_3 nitrogen trichloride
12. SiO_2 silicon dioxide
13. PCl_3 phosphorus trichloride
14. XeF_4 xenon tetrafluoride
15. P_4O_{10} tetraphosphorus decoxide

Write the formula of the molecular compound from the given name.

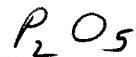
1. Silicon Dioxide



2. Sulfur Hexafluoride



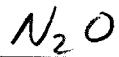
3. Diphosphorus Pentoxide



4. Xenon Tetrafluoride



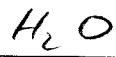
5. Dinitrogen Monoxide



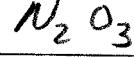
6. Carbon Monoxide



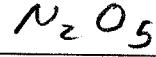
7. Dihydrogen Monoxide



8. Dinitrogen Trioxide



9. Dinitrogen Pentoxide



10. Dinitrogen Tetraoxide

